

MBZ-003-1102001 Seat No. _____

M. Sc. (Sem. II) (CBCS) Examination

April / May - 2018

C-201: Inorganic Chemistry

Faculty Code: 003

Subject Code: 1102001

Time: $2\frac{1}{2}$ Hours] [Total Marks: 70]

Instructions: (1) All Questions are compulsory.

(2) All Questions carry equal Marks.

1 Answer the following (Any Seven)

14

- (a) Define σ -bonded OMC
- (b) Discuss the limitations of ESR spectroscopy
- (c) Draw the structure of Zeise's Salt
- (d) Discuss the work of Hemoglobin in our body
- (e) Discuss the use of Ion exchange chromatography
- (f) Discuss the role of Tannin in analysis
- (g) Draw the ESR spectrum when one electron influenced by a single proton of the
- (h) Discuss Physiology of blood
- (i) Name some of the most important Ion exchange resins
- (j) Give the deference between isotropic 'g' value and anisotropic 'g' value in ESR spectroscopy
- 2 Answer the following : (Any Two)

14

- (a) Discuss the preparative methods of η^3 allyl OMC of transition metals
- (b) Give the classification and role of metal ions according to their action in Biological System

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(c) Write short note on ESR instrumentation

3	Answer the following: (Any Two)		14
	(a)	What is Hyperfine spitting in ESR spectroscopy	
	(b)	Write note on the role of Iodine in activity of Thyroid hormones	
	(c)	Give the classification of π –boned OMC of transition metals	
4	Ans	ewer the following:	14
	(a)	Discuss the preparation and use of following reagents in Inorganic analysis	
		(1) DMG	
		(2) Salicylaldoxime	
	(b)	Discuss the ESR spectrum of of H_2^+ .	
5	Answer the following:		14
	(a)	Describe toxic elements, toxicity and deficiency with suitable example	
	(b)	Discuss the physical properties of πbonded OMC of transition metals	
		\mathbf{OR}	
5	Answer the following:		14
	(a)	Discuss Metalloporphyrins in detail	
	(b)	Define Ion Exchange Chromatography and its use in separation of Cadmium and Zinc.	